

## Test I (28.07.04) : Review of Thermodynamics

1. - Ask yourself a question. [5]
2. - Answer the question. [5]

## Test II (24.08.04) : Ensemble Formulation

Consider a thermodynamics system, in which  $E, N, V$  are allowed to vary. Write down (with justification) the formal expression for,  $Z_s$ , the super-canonical partition function, suitable for describing this system. Express,  $\langle V \rangle$  and  $\langle \delta V \rangle^2$  as appropriate derivatives of  $Z_s$ . Give two examples of such a system. [9+1]

## Test III (14.09.04) : Non-Interacting Particles

**A three level system** - Consider a system of  $N$  particles, confined within a volume  $V$ . The single particle energy levels consist of one ground state of zero energy and two degenerate excited states of energy  $\varepsilon$ .

1. Calculate the canonical partition function.
2. Calculate the probability that the (single particle) system is its ground state. Plot this as a function of  $T$  and discuss the limiting cases.
3. Calculate the average energy and entropy. Plot the average energy per particle and entropy per particle as functions of  $T$ .