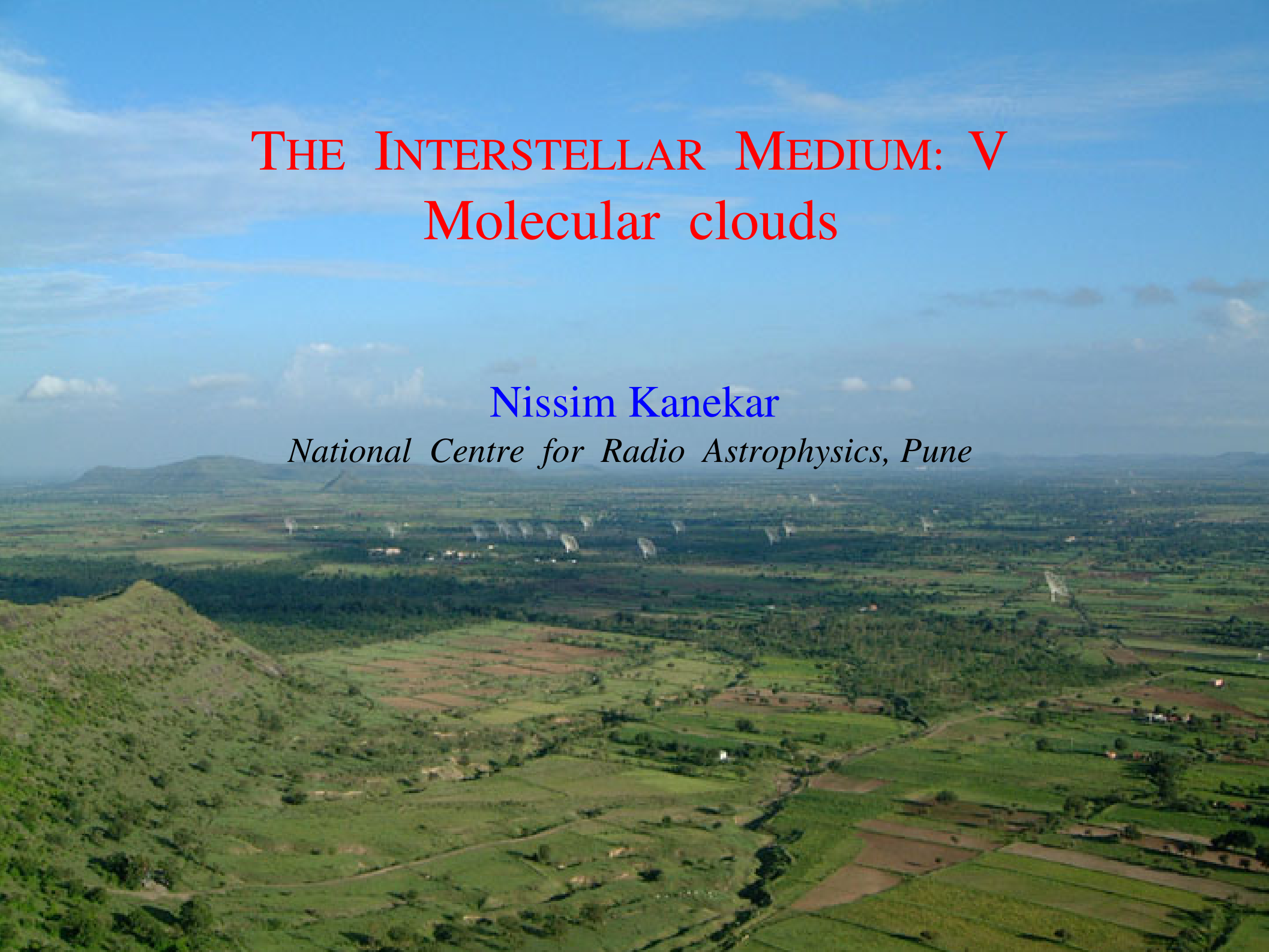


THE INTERSTELLAR MEDIUM: V

Molecular clouds

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OUTLINE

- Background.
- Interesting atomic lines: The [CII] 609/370- μm lines.
- Molecular transitions: electronic, rotational, vibrational...
- Molecular hydrogen: The electronic Lyman and Werner bands.
- Conditions in diffuse clouds.

BACKGROUND

- HI-21cm studies: Estimates of N_{HI} , T_{K} , T_{S} , B_{\parallel} , scale height.
External galaxies: Mass, spatial distribution, velocity field,
- Collisional & radiative excitation / de-excitation: (2-level atom)
$$(n_1/n_0) = [n_c k_{01} + n_\gamma (g_1/g_0) A_{10}] / [n_c k_{10} + (1 + n_\gamma) A_{10}]$$
- Critical density: $n_{crit,u} = [\sum_{l<u} (1 + n_\gamma) A_{ul}] / [\sum_{l<u} k_{ul}]$
If $n_c \gg n_{crit,u} \Rightarrow T_X = T_K$; if $n_c \ll n_{crit,u} \Rightarrow T_X = T_R$.
- [CII]-158 μm line: “Forbidden” transition; $A_{10} \sim 2.4 \times 10^{-6} \text{ s}^{-1}$.
- $n_{crit} \sim 3000 \text{ cm}^{-3} \gg$ Number density in diffuse clouds $\Rightarrow T_X \ll T_K!$
- Collisional excitation, radiative de-excitation \Rightarrow Radiative losses!
- [CII]-158 μm line most important cooling route in diffuse gas!
 $\sim 0.5\%$ of a galaxy's luminosity! Very important line for ALMA!

THREE LEVEL ATOMS: CI

- Three level system: More complex!

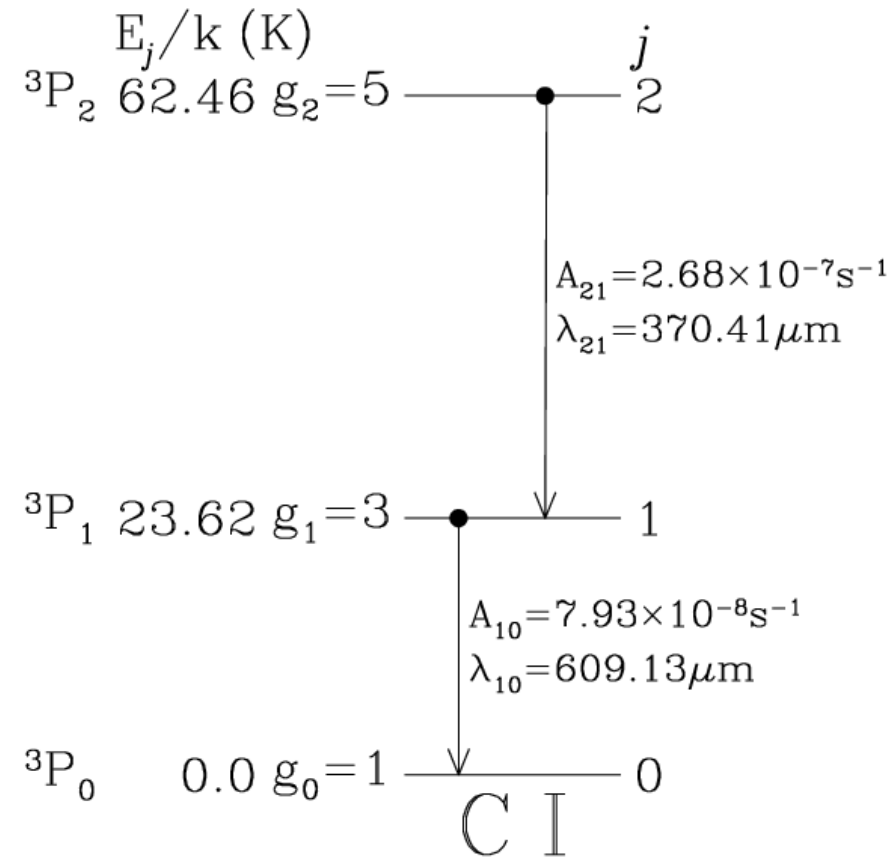
- $(dn_2/dt) = R_{02}n_0 + R_{12}n_1 - (R_{20} + R_{21})n_2$

- $(dn_1/dt) = R_{01}n_0 + R_{21}n_2 - (R_{10} + R_{12})n_1$

- E.g. $R_{10} = C_{10} + (1 + n_{\gamma,10})A_{10}$;

- $R_{01} = (g_1/g_0)[C_{10}e^{-E_{10}/kT} + n_{\gamma,10}A_{10}]...$

- In steady state, $(dn_2/dt) = (dn_1/dt) = 0$.



(Draine 2011)

- $(n_1/n_0) = (R_{01}R_{20} + R_{01}R_{21} + R_{21}R_{02}) / (R_{10}R_{20} + R_{10}R_{21} + R_{12}R_{20})$

- $(n_2/n_0) = (R_{02}R_{10} + R_{02}R_{12} + R_{12}R_{01}) / (R_{10}R_{20} + R_{10}R_{21} + R_{12}R_{20})$.

THREE LEVEL ATOMS: CI

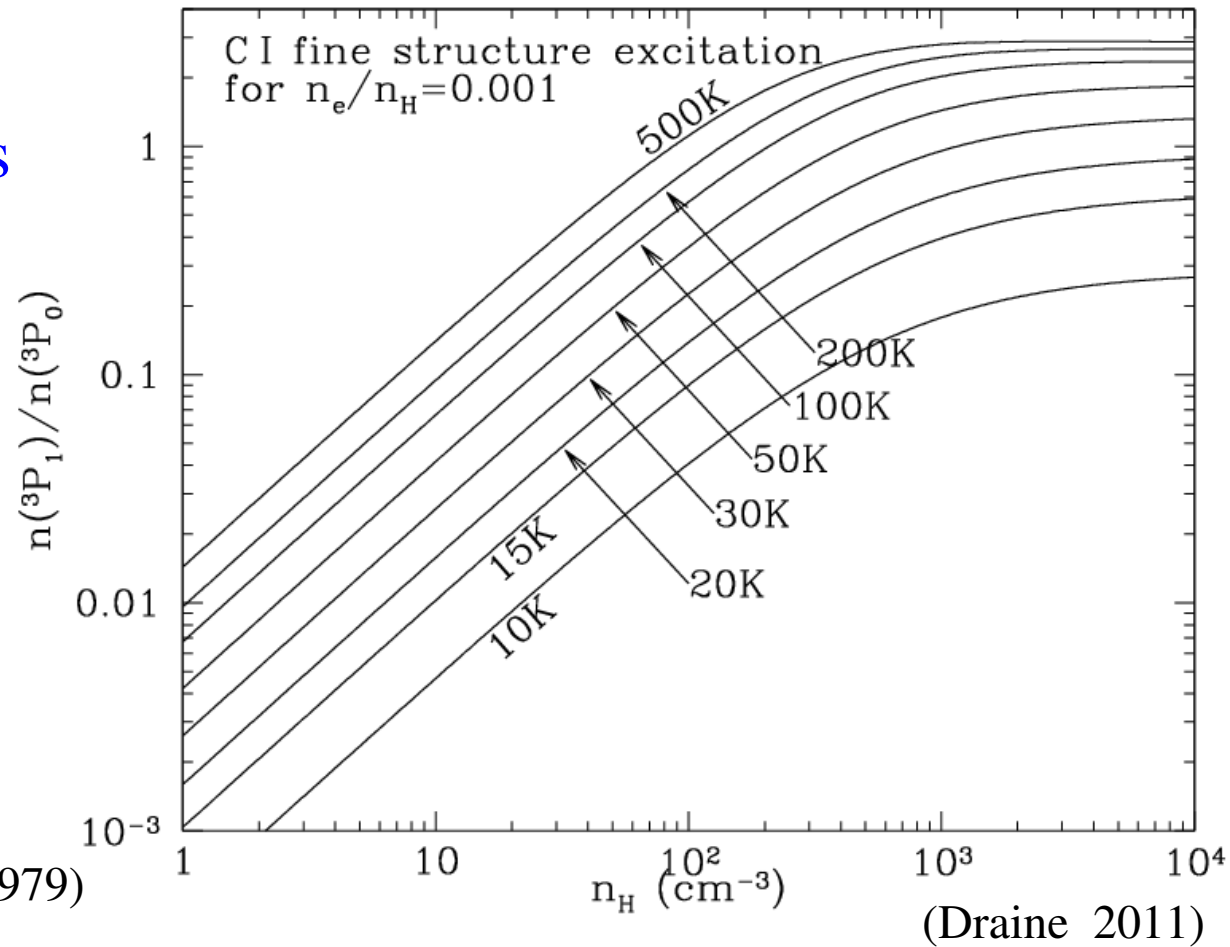
- $n_{crit} \sim 700 \text{ cm}^{-3}$ ($T_K \sim 100 \text{ K}$) ; $n_{crit} \sim 150 \text{ cm}^{-3}$ ($T_K \sim 5000 \text{ K}$).
Densities in diffuse ISM much lower than critical density!

- Since $n < n_{crit}$, the level populations depend on n & T_K !

- Level populations obtained from strengths of UV lines to higher excited states.

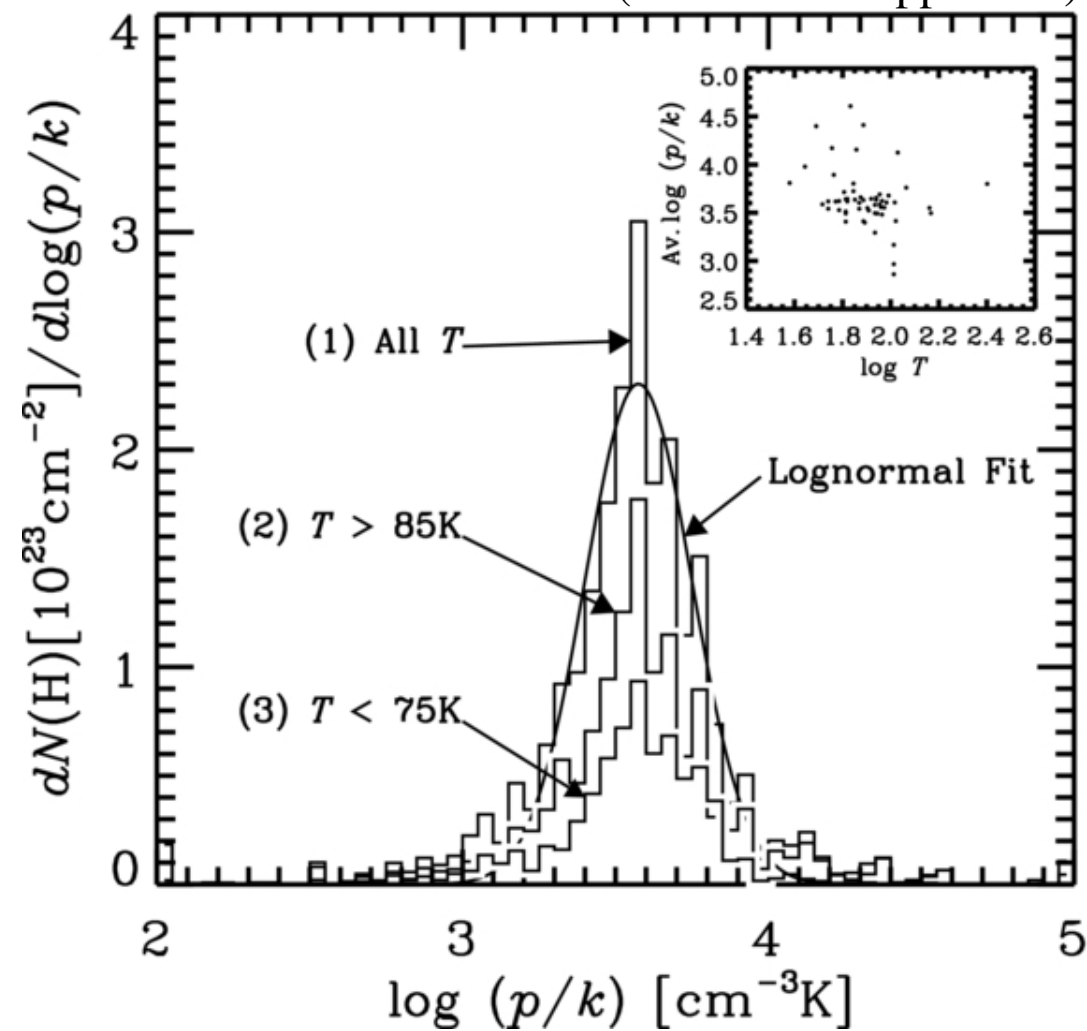
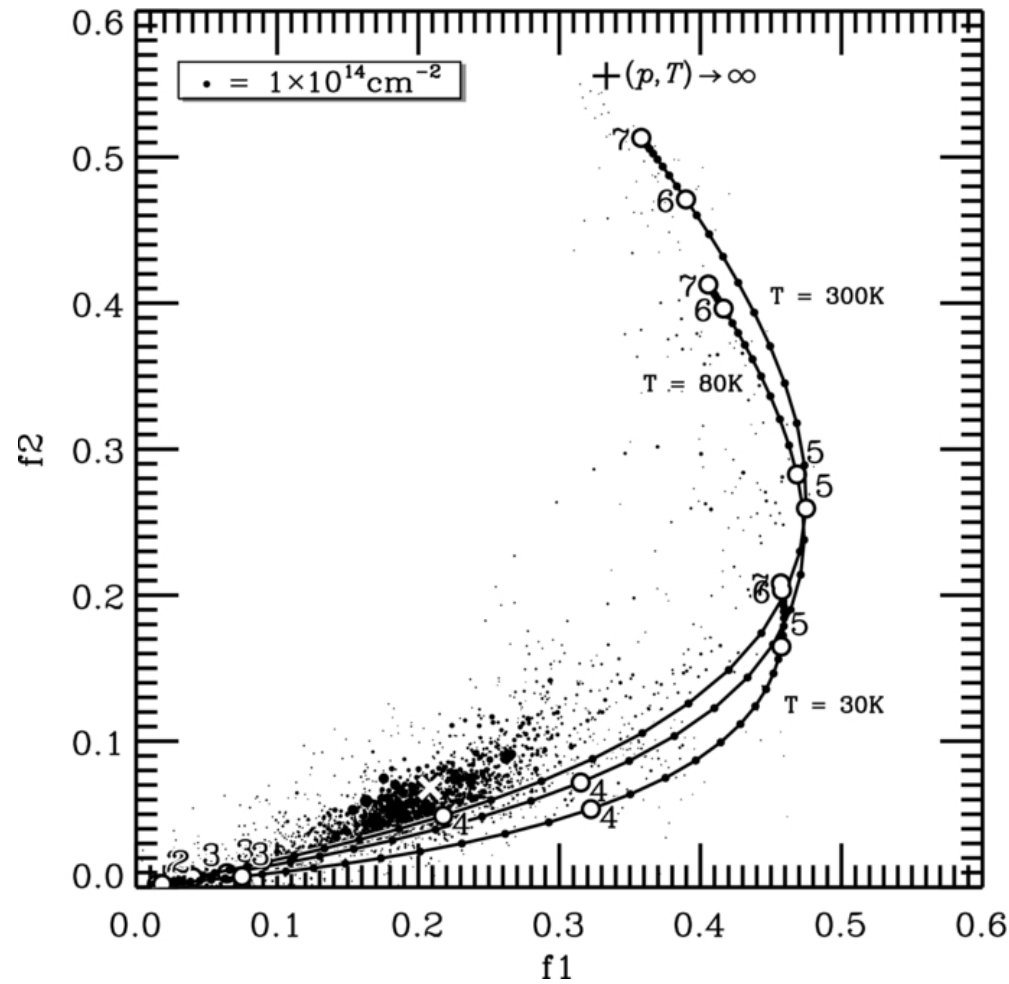
- Three levels \Rightarrow Solve for density and temperature!
 \Rightarrow Infer thermal pressure!

(Jenkins & Shaya 1979)



THREE LEVEL ATOMS: CI

(Jenkins & Tripp 2011)



• Most points lie above and to the left of the curves!

⇒ Two pressures, $3800 \text{ cm}^{-3} \text{ K}$ and $> 10^4 \text{ cm}^{-3} \text{ K}$???

(Jenkins & Tripp 2011)

MOLECULAR LINES: VIBRATION

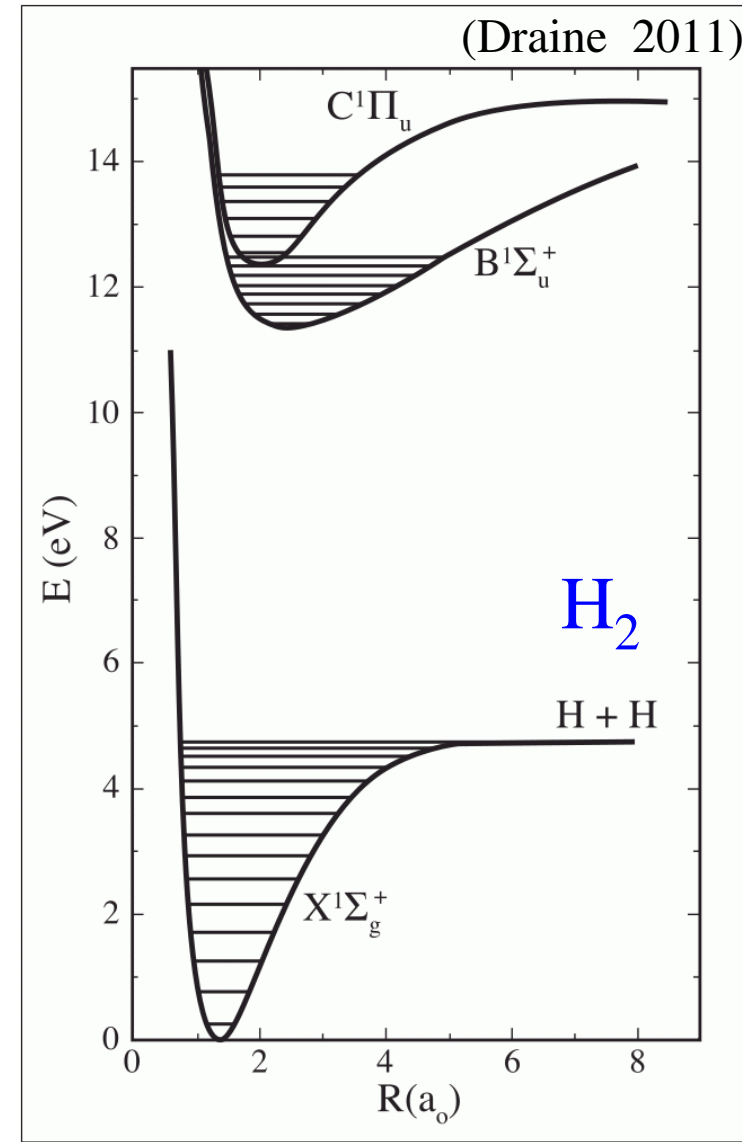
- Diatomic molecules: Stretch along the line joining the nuclei.

- Vibration: Assume fixed internuclear separation, solve for electron ψ_q & E_q . For slow nuclear motion, ψ_q & E_q will change adiabatically.
 \Rightarrow Nuclear motion under effective potential $V_q(r_n) = E_q(r_n) + (Z_1 Z_2 e^2 / r_n)$.

- Near minimum r_0 , $V_q(r) = V_0 + (1/2)k(r-r_0)^2$
Harmonic oscillator: $\omega_0 = (k/m_r)^{1/2}$.

- Vibrational modes: $E = h\nu_0(v + 1/2)$

Low mass \Rightarrow High vibration frequency (e.g. H_2 : $2.1\mu\text{m}$).



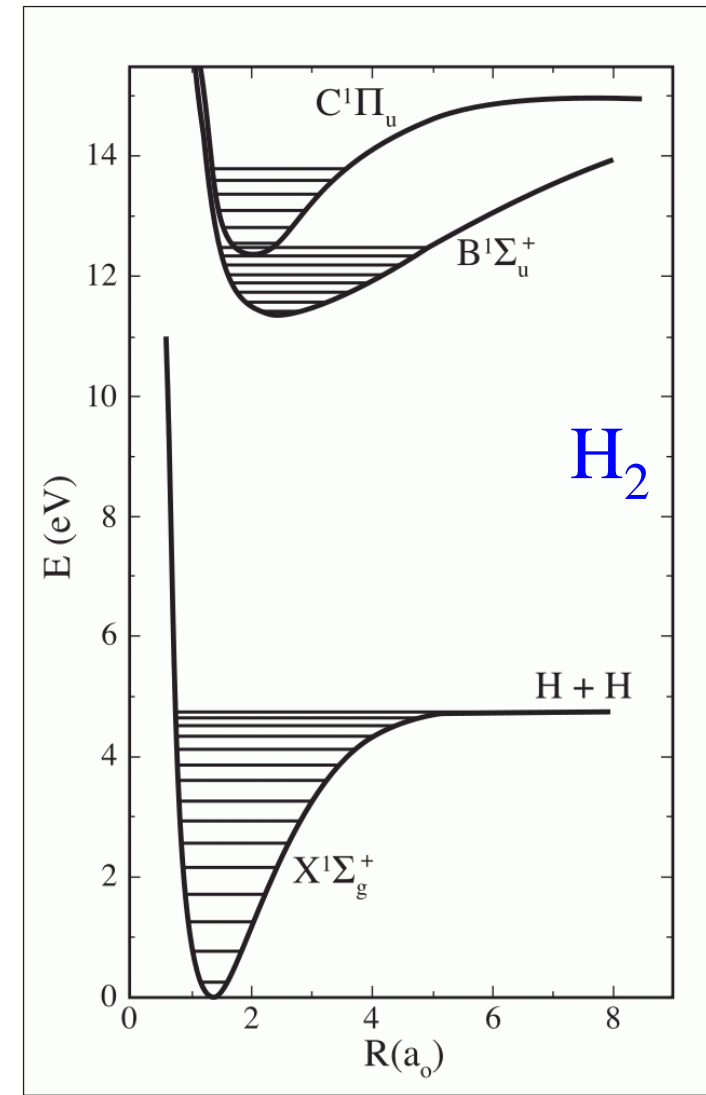
MOLECULAR LINES: ROTATION

- Diatomic molecules: Rotate around axis \perp to molecular axis.
- Classical kinetic energy of a rigid rotor = $(J\hbar)^2/2I$,
where $I = m_r r_0^2$ is the moment of inertia.
 \Rightarrow Quantum mechanical rotational energy = $J(J+1)\hbar^2/2m_r r_0^2$.
- Total vibration-rotation energy in electronic state q :
$$E_q = V_q(r_0) + h\nu_0 (v + 1/2) + B_v J(J+1)$$
with $\nu_0 = (\omega_0/2\pi) = (1/2\pi)(k/m_r)^{1/2}$ & $B_v = \hbar^2/2m_r r_0^2$.
Vibration lines, $v \rightarrow v-1$: $h\nu_0$; Rotation lines $J \rightarrow J-1$: $2B_v J$
- None of the above is *exact!!!* Not a quadratic potential, nor a rigid rotor; I depends on vibrational & rotational levels.
- *All* molecules have rotation/vibration states, besides electronic states.

WAVELENGTHS & SELECTION RULES

(Draine 2011)

- Electronic transitions : UV or optical.
Pure vibrational transitions : Mid-IR.
Pure rotational transitions : Radio – far-IR.
- Rotational : (1) Permanent dipole moment.
(2) $\Delta J = \pm 1$.
- Vibrational: (1) $\Delta v = \pm 1$.
(2) $\Delta J = \pm 1$ (for Σ states).
- Electronic : Complex; rules different for different symmetries.



- Homonuclear molecules (e.g. H₂): No permanent dipole moment!
⇒ Rotational and vibrational transitions “forbidden”.

MOLECULAR HYDROGEN, H₂

- “para-H₂”: (S=0, J=0,2,4,...) & “ortho-H₂” (S=1, J=1,3,5,...). $\Delta J = \pm 2$.
- Symmetric molecule \Rightarrow No permanent electric dipole moment!
 \Rightarrow Rotational and vibrational transitions “forbidden”.
- Electric quadrupole rotation/vibration lines: Low Einstein A's!
- Low mass \Rightarrow High vibration, rotation frequency ($\sim 2.1, 28 \mu\text{m}$).
 \Rightarrow Upper states only excited at high temperatures ($T_* \sim 510 \text{ K}$)
 \Rightarrow No detectable emission from normal molecular clouds!
- Electronic lines: Only direct tracer of H₂ in molecular clouds!
Typically detected in absorption against stars/quasars.
- **But...** Far-UV wavelengths ($\lambda < 1110 \text{ \AA}$) \Rightarrow Only detectable on diffuse sightlines without much dust extinction (i.e. low N_{HI}).

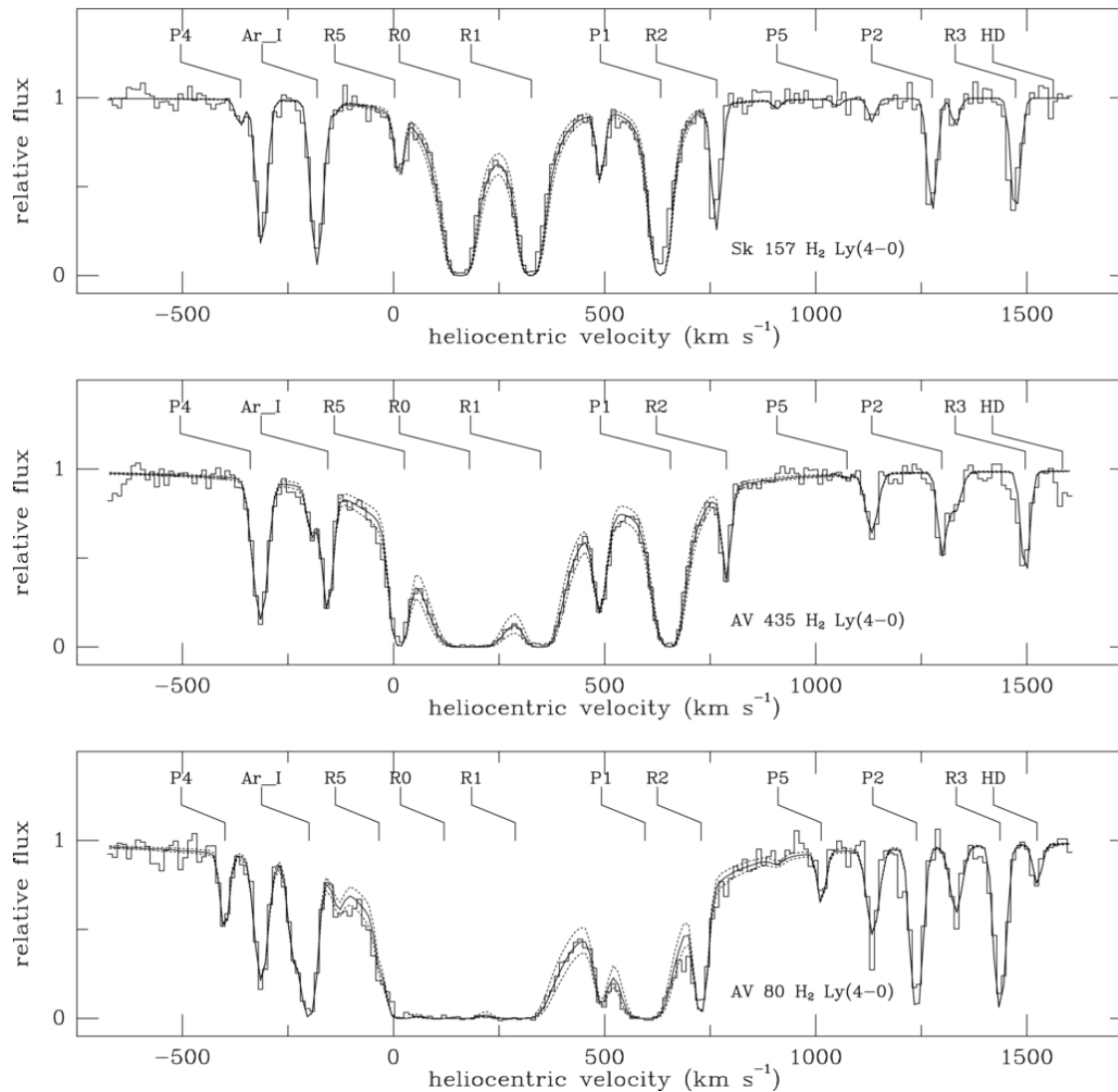
MOLECULAR HYDROGEN, H₂

- Lyman ($\lambda < 1110 \text{ \AA}$) and Werner bands: $\lambda < 1021 \text{ \AA}$.

⇒ Mostly studies with the *Copernicus* and *FUSE* satellites.

(e.g. Savage et al. 1977;
Rachford et al. 2009)

H₂ in the LMC/SMC



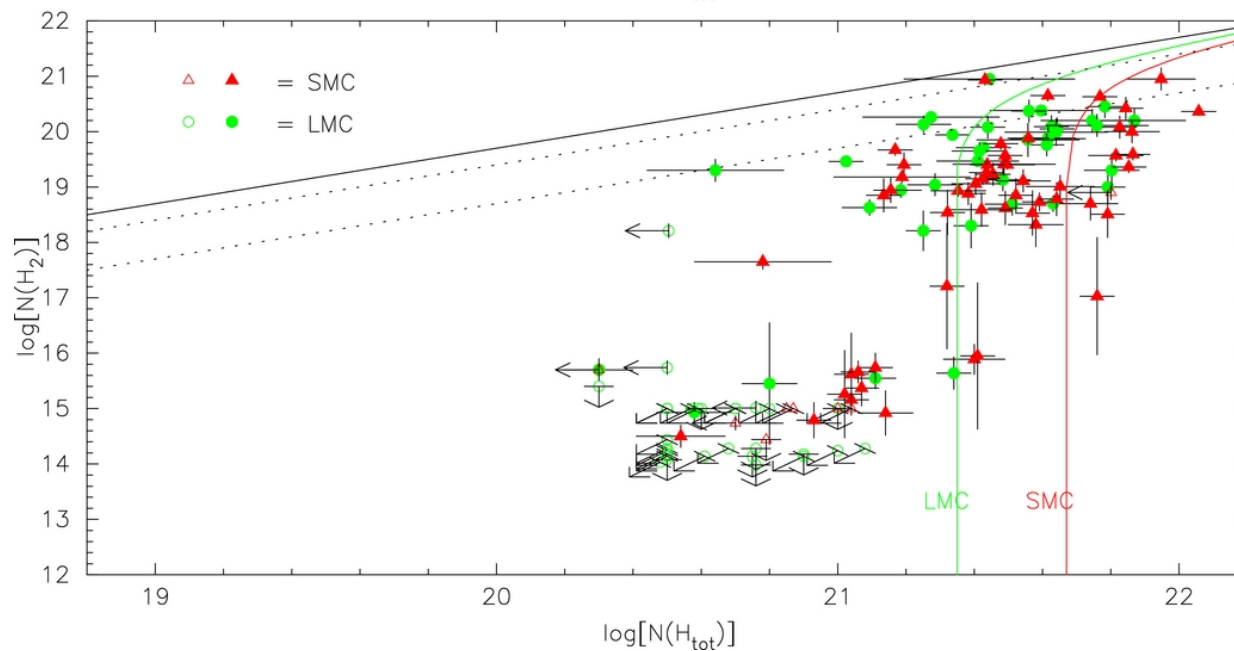
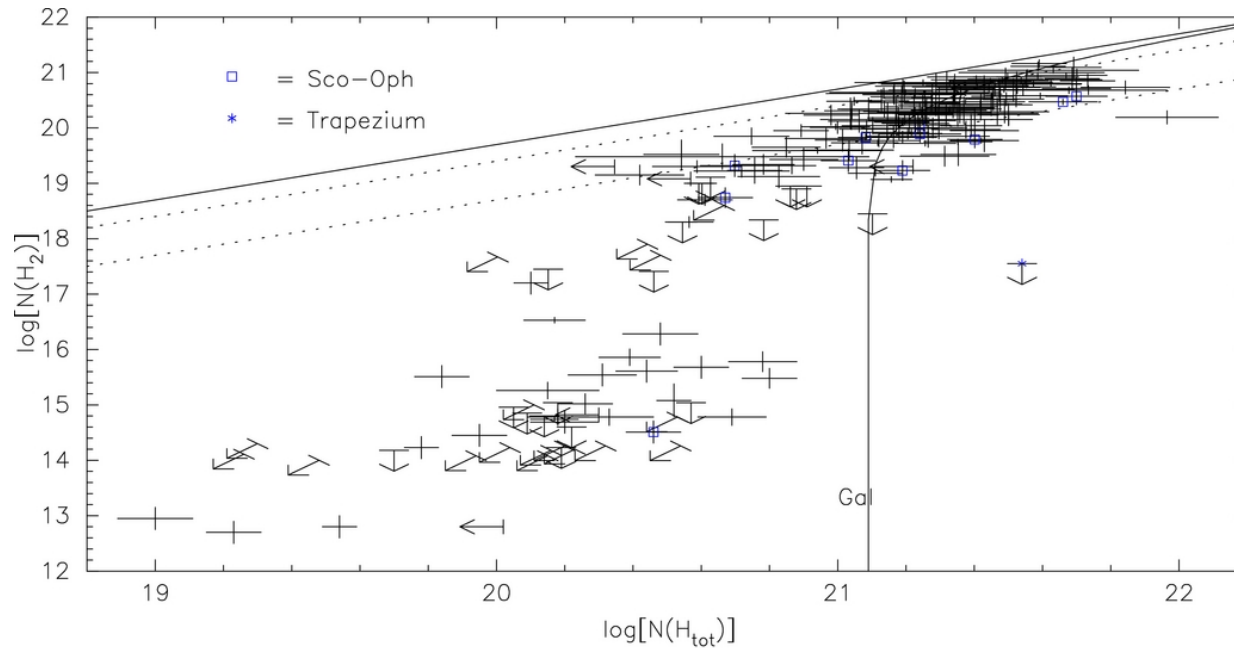
(Welty et al. 2012)

MOLECULAR HYDROGEN, H₂

- Lyman/Werner absorption studies \Rightarrow Level populations in ro-vibrational levels of ground electronic state & total N_{H_2} .
- J=1 and J=0 levels thermalized by collisions $\Rightarrow T_{\text{ROT}} = T_{\text{K}}$!
 $\Rightarrow T_{\text{K}} \sim 77 \text{ K}$ in cold diffuse gas. (Savage et al. 1977)
- Correlation between total H column density ($N_{\text{H}} = N_{\text{HI}} + 2N_{\text{H}_2}$) and “reddening” $E_{\text{B-V}} = A_{\text{B}} - A_{\text{V}}$ (B: 4405 Å; V: 5470 Å).
$$N_{\text{H}} = 5.8 \times 10^{21} E_{\text{B-V}} \text{ cm}^{-2} \text{ mag}^{-1}.$$
 (Bohlin et al. 1978; Rachford et al. 2009)
- Constant higher by a factor of ~ 3 (LMC) and ~ 5 (SMC), similar to metallicity factors in LMC and SMC. (Welty et al. 2012)

MOLECULAR HYDROGEN, H₂

- Metallicity-dependent N_{HI} threshold for H₂ formation.



(Welty et al. 2012)

DIFFUSE CLOUDS: PROPERTIES

- Multi-phase medium: CNM with $T_K \sim 70$ K and high density.
WNM with $T_K > 1000$ K and low density.
- Extended far beyond the stars in normal galaxies.
Scale heights of ~ 75 pc (CNM) and ~ 220 pc (WNM).
- $T_S \sim T_K$ in the CNM; $T_S < T_K$ in the WNM.
- Magnetic field strengths of ~ 5 μ G in the CNM.
- Pressures of ~ 3800 cm^{-3} K, from CI absorption. **But...**
- Important coolant: [CII]-158 μ m line.
- Cold phase formation at $N_{\text{HI}} \sim 2 \times 10^{20}$ cm^{-2} in the Galaxy.
H₂ formation at $N_{\text{HI}} \sim 6 \times 10^{20}$ cm^{-2} .
Metallicity-dependent threshold! Higher threshold in LMC/SMC !